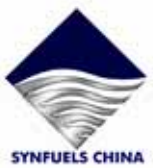


# **Fischer-Tropsch synthesis from laboratory to engineering scale up**

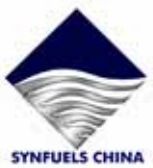
**Hao Xu**

**Institute of Coal Chemistry, Chinese Academy of  
Sciences  
State key laboratory of Coal chemistry  
Synfuels China®**



# Introduction :

- **Government need**
- **International technology situations**
  - Sasol: high temperature FTS**
    - Low temperature FTS**
  - Shell: SMDS**
- **Technology development in China**
  - Synfuels China® : HFPT ICC-IA**
    - LFPT ICC-IIA**



# **CTL process:**

## **Coal gasification**

**Fixed-bed or moving bed gasifier**

**Fluid-bed gasifier**

**Entrained flow gasifier**

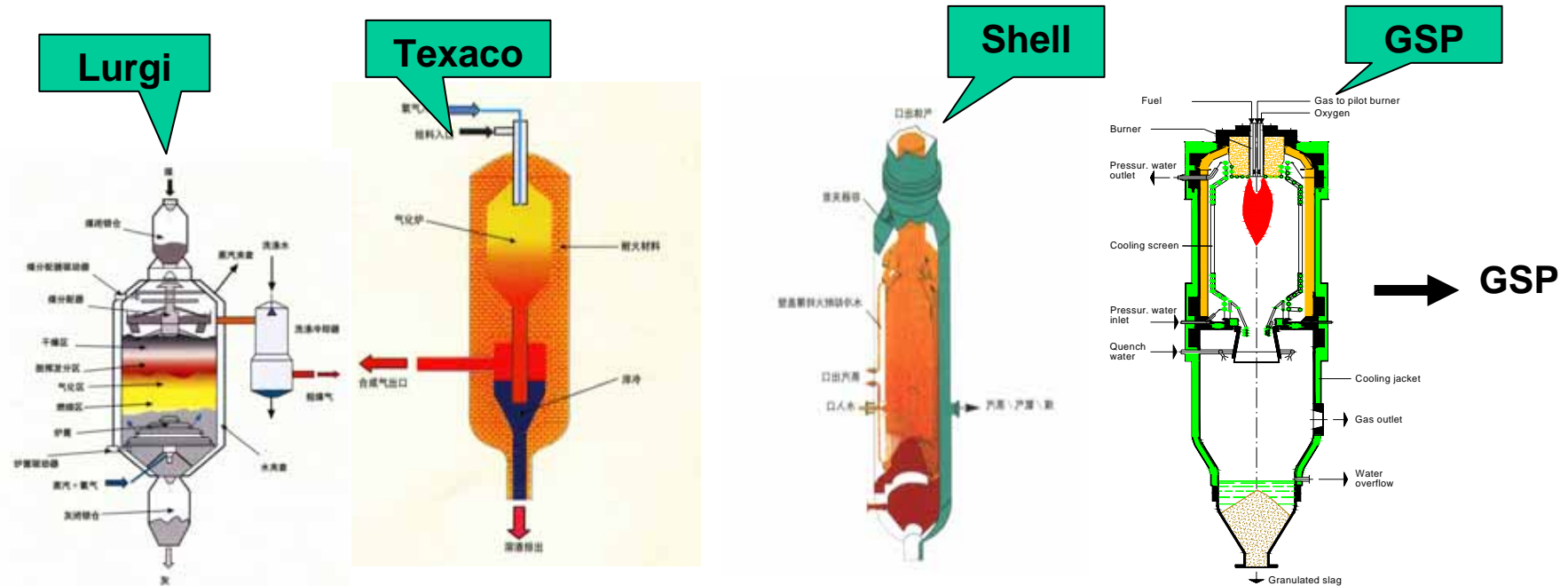
**shift reaction**

**Acidic gas removal**

**Slurry FTS**

**Product upgrading**

# Coal gasification



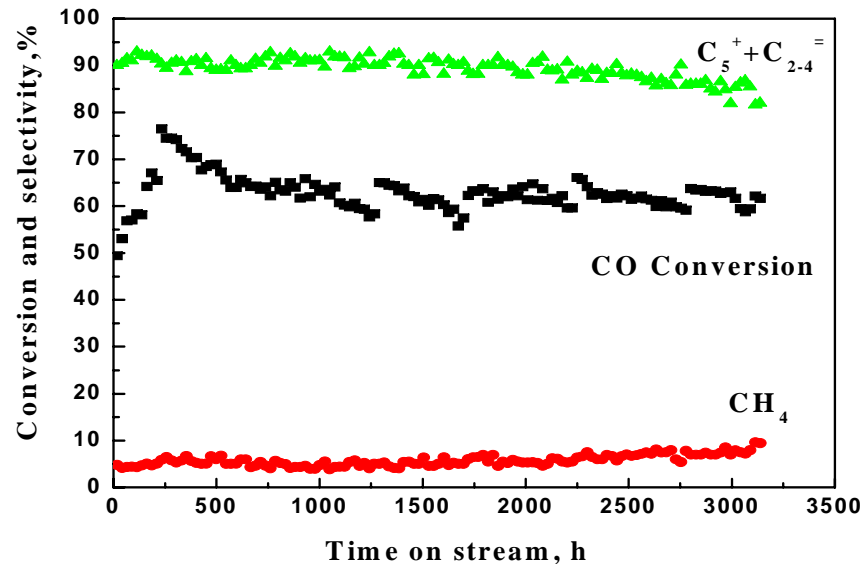
Cheap in construction  
Methane rich syngas  
Coal adaptability  
Tar recovery  
Water treatment

Expensive in construction  
High quality Syngas  
Coal adaptability \*  
Short time on stream  
High O<sub>2</sub> consumption

Expensive in construction  
High quality Syngas  
Coal adaptability\*\*  
Hot stream quenching  
Large boiler size

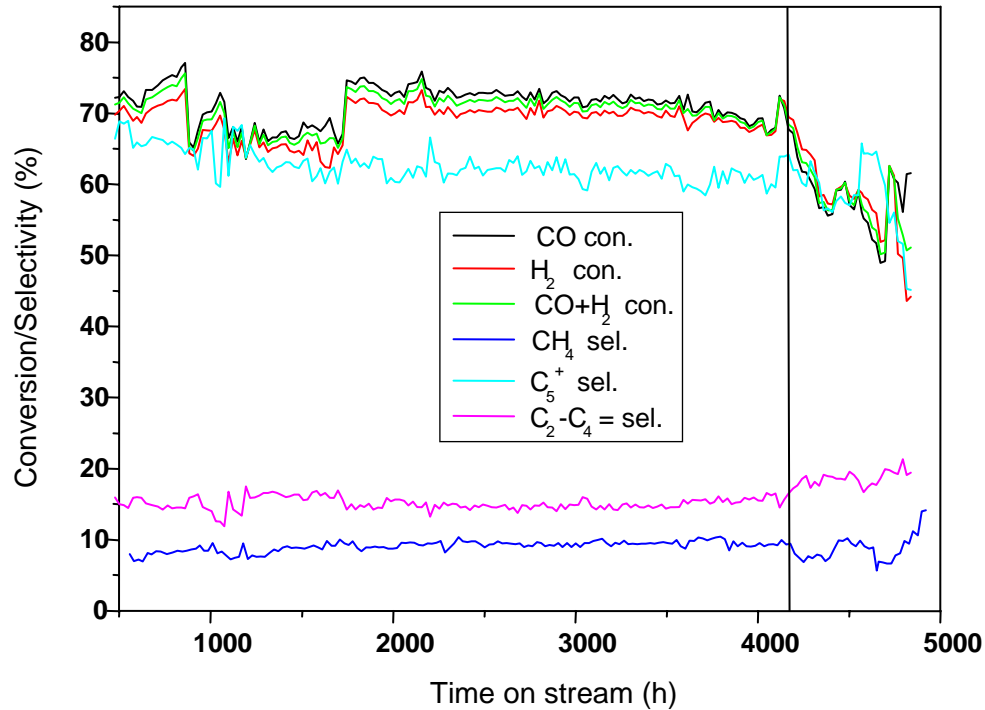
Cheap in construction  
High quality Syngas  
Coal adaptability\*\*\*  
Quenching type and heat recovery  
BDI SVP  
FUTURE ENERGY  
CHOREN

# Laboratory research : catalyst development and research ICC-IA catalyst



Changes of conversions and selectivity with time-on-stream in slurry phase reactor (240 -270 °C, 1.5-2.0 MPa, 2.0 NL/g-cat./h, H<sub>2</sub>/CO=0.67)

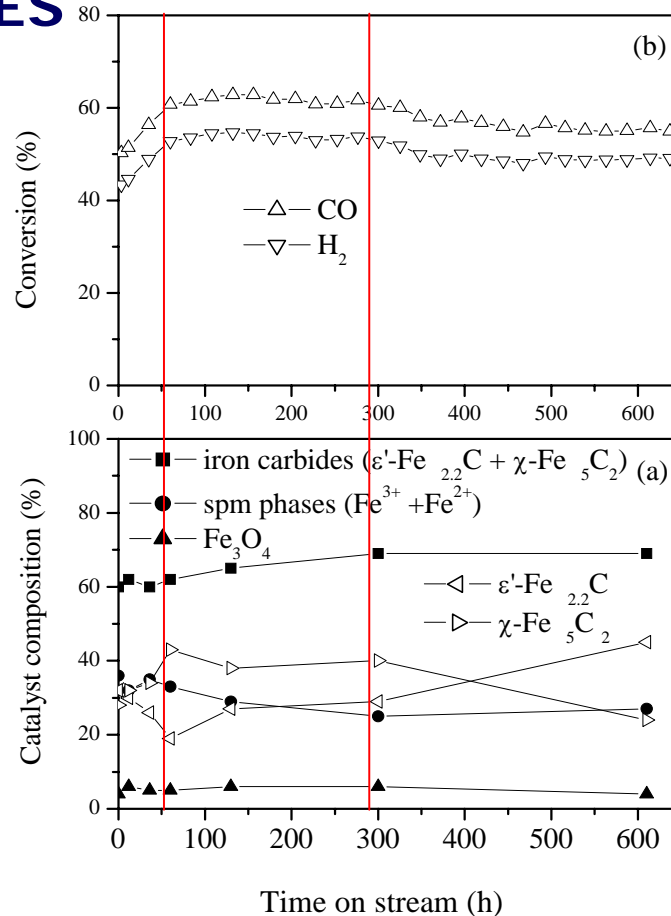
# Laboratory research : ICC-IIA catalyst



Changes of conversions and selectivity with time-on-stream in slurry phase reactor (240 -270 °C, 1.5-2.0 MPa, 2.0 NL/g-cat./h, H<sub>2</sub>/CO=0.67)

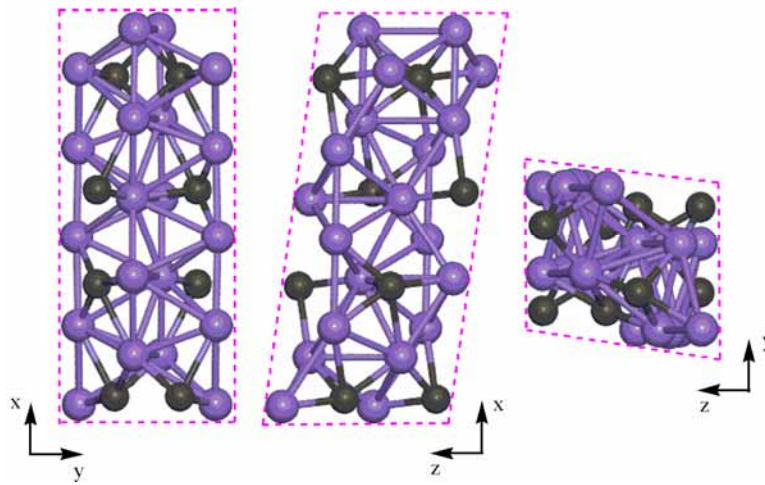
# Catalyst characterization

## XRD, SEM, MES



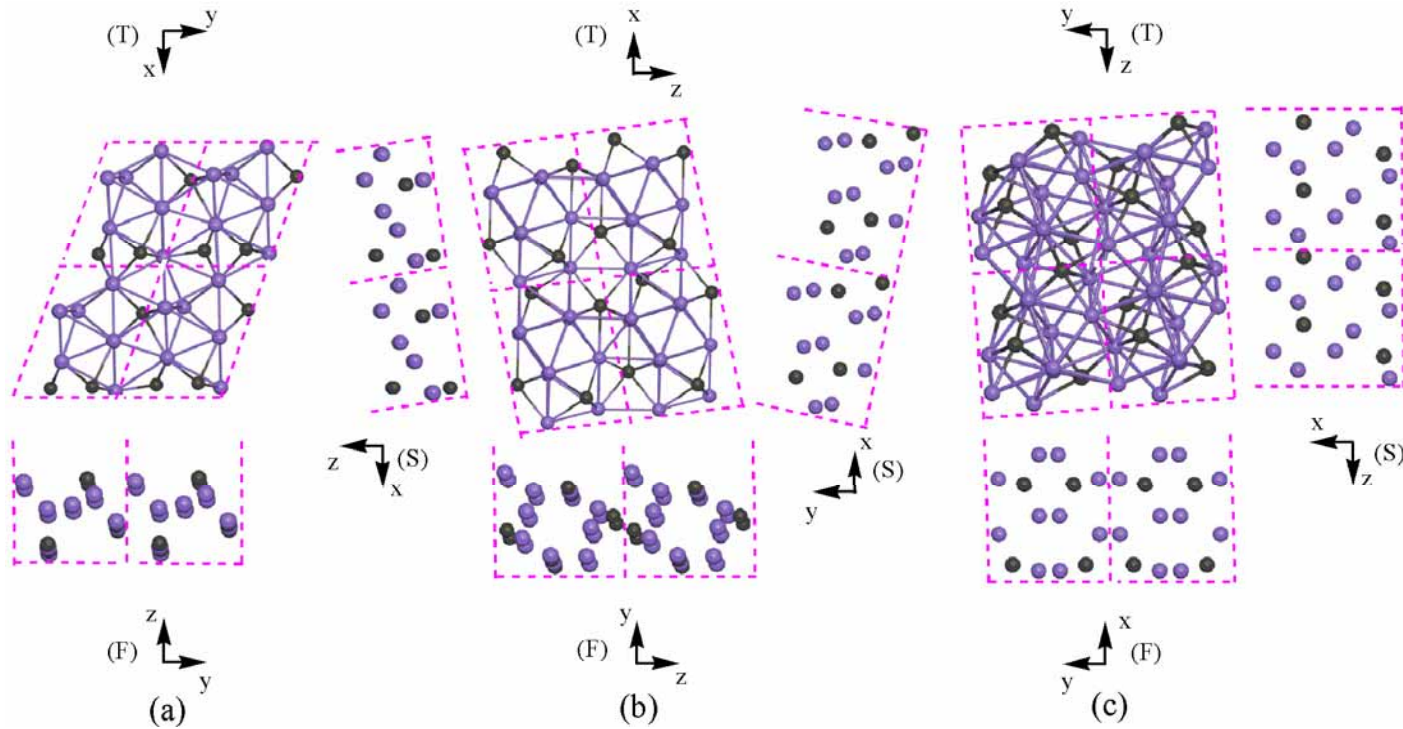
Correlative studies of phase transfer and activities of an ICC-IA model catalyst during FTS reaction.  
(240 -270 , 1.5-2.0 MPa, 2.0 NL/g-cat./h, H<sub>2</sub>/CO=0.67)

# Quantum chemistry calculation

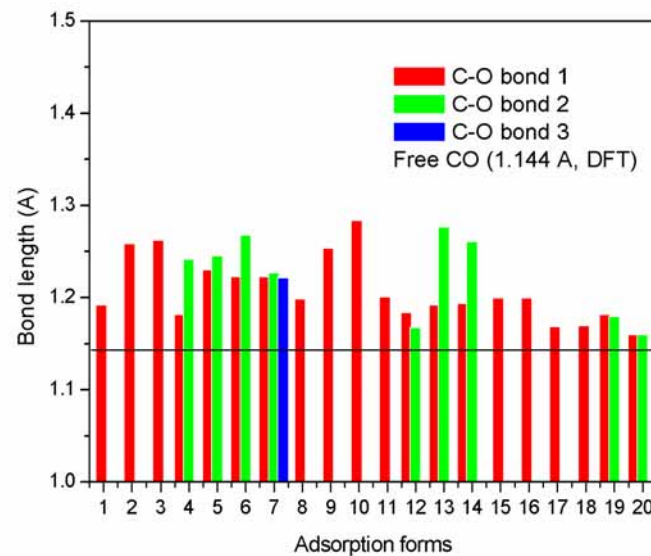
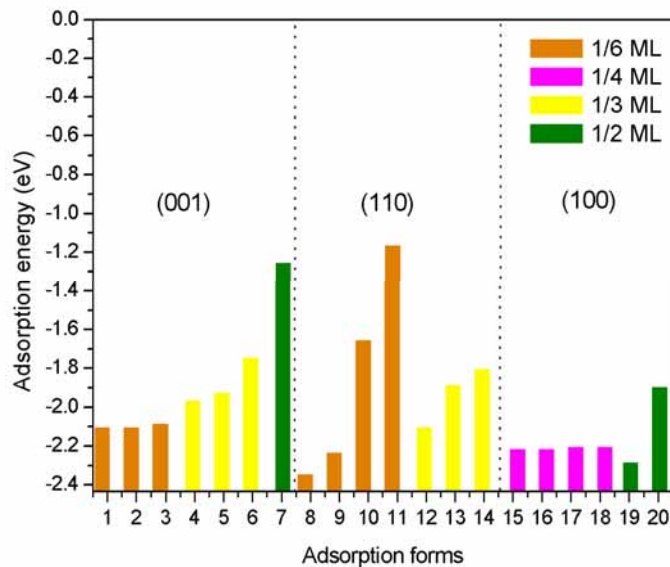


**Schematic unit cell of Fe<sub>5</sub>C<sub>2</sub>. (Fe: light gray; C: deep gray.)**  
**(1) CO adsorption on the surface of Fe<sub>5</sub>C<sub>2</sub> (001)、(110)、(100)**





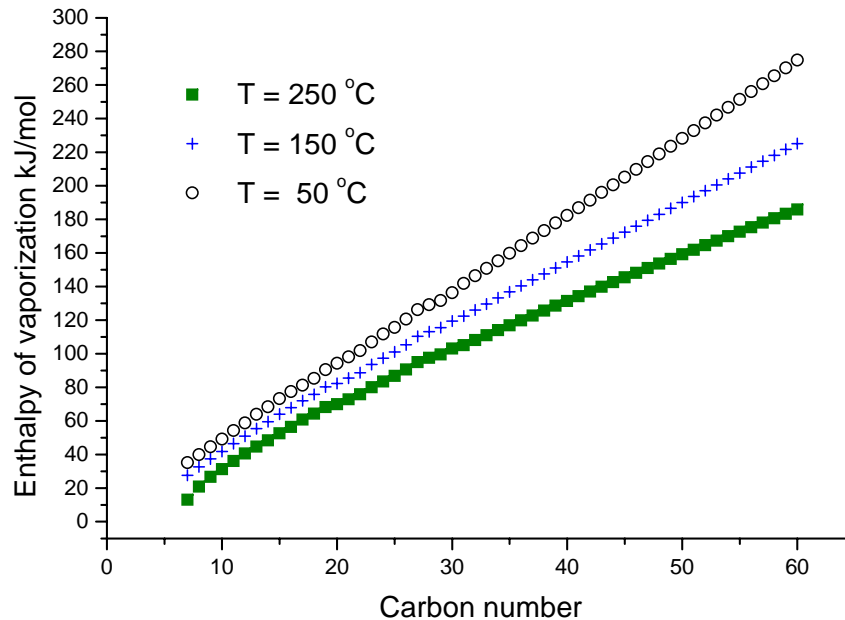
**Schematic top and front views of Fe<sub>5</sub>C<sub>2</sub>(001) (a); top and side views of (110) (b); and top and front views of (100) (c) in a p(1 × 1) unit cell.**



Adsorption energetics (left) and C-O bond activation (right) of CO adsorption on Fe<sub>5</sub>C<sub>2</sub> surfaces

# Kinetic and slurry reactor model research

## Develop user databank for FTS



Enthalpy of vaporization at different temperature for n-paraffins, C7 ~ C30 data are from Aspen Plus, C31 - C60 data are from this work

# FTS Kinetic research :

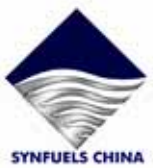
Kinetic research for Fe-Cu-K catalyst

Ma Wen-pin's model

Wang Yi-ning's Model

Zhang rong-le's model





# **FTS Kinetic research :**

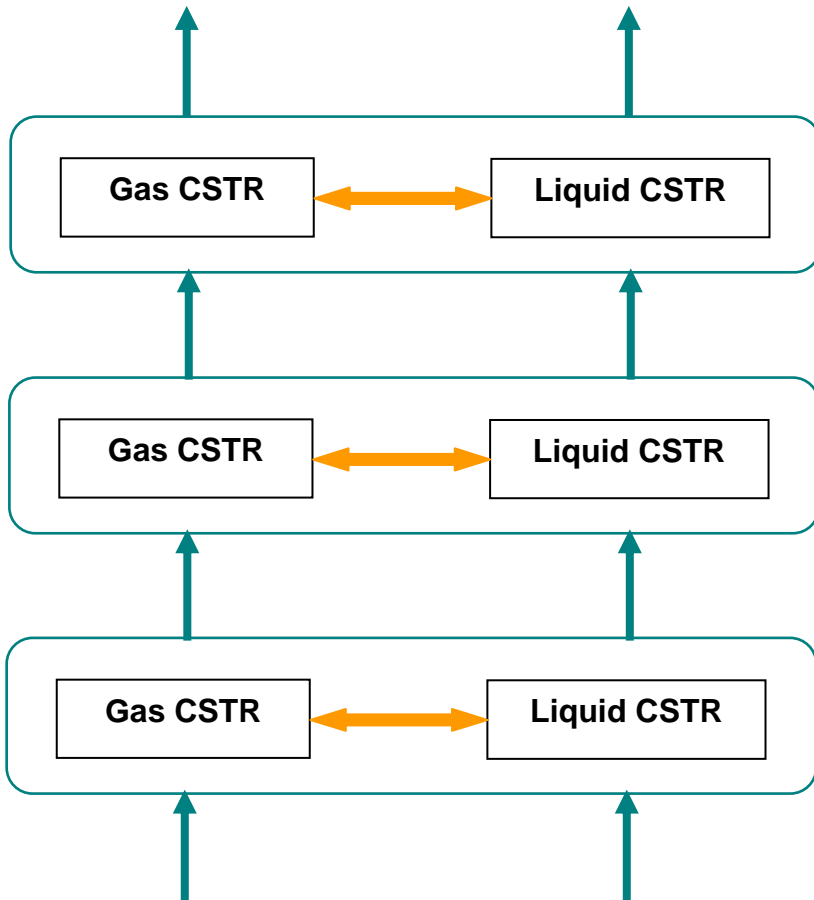
## **Kinetic research for Fe-Cu-K catalyst**

**Ji Yuan-yuan's model**

**Yang Jun's Model**

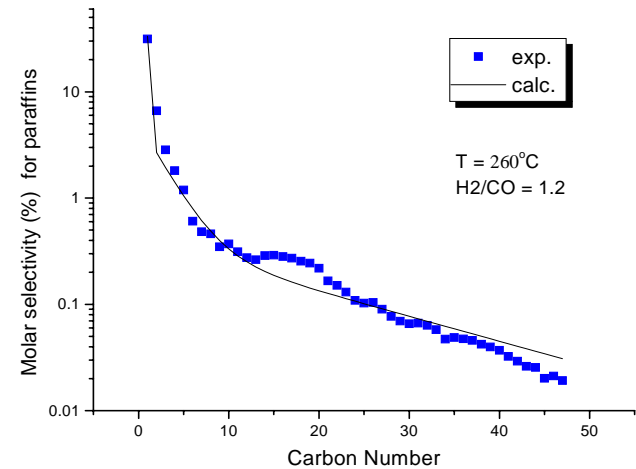
**Teng Bo-tao's model**

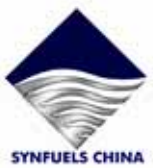
# Develop slurry bubble reactor



$$\varepsilon_g \frac{dC_{g,i}}{dt} = \frac{u_{g0}C_{g0,i} - u_g C_{g,i}}{h} - (k_{L,i} a) \left( \frac{C_{g,i}}{m_i} - C_{L,i} \right)$$

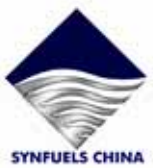
$$\varepsilon_L \frac{dC_{L,i}}{dt} = \frac{u_{L0}C_{L0,i} - u_L C_{L,i}}{h} + (k_{L,i} a) \left( \frac{C_{g,i}}{m_i} - C_{L,i} \right) - \frac{W_{cat}}{hS} \sum_j \alpha_{i,j} R_j$$





## Pilot plant research

- Construction began in 2001
- First run in 2002, 25 days
- Modifications in winter of 2002
- Three runs in 2003,
- Two runs in 2004, 1500 hours
- Catalyst, kinetic model and slurry bubble reactor were tested in the pilot plant.



# Develop product upgrading technology

- Cooperate with industrial partner to Develop hydrogenation and hydrocracking technology
- Developing hydrogenation and hydrocracking technology in laboratory



## CTL diesel properties (Synfuels China®)

Indexes	Hydrogenated Fraction Refinery Co-MoS <sub>2</sub> catalyst	Hydrocracked Co-MoS <sub>2</sub> catalyst	Hydrogenated Fraction S-free catalyst
Density kg/L	0.7684	0.7782	0.7664
Viscosity /mm <sup>2</sup> /s, 20 °C	3.172	3.957	3.276
Initial boiling point /°C	150	130	185
Final boiling point /°C	370	385	330
Disti. Range /°C			
5%			185-206.2
10%	154-188	138-184	213.7
30%	257	231	232.7
50%	278	272	252.7
70%	293	305	278.2
90%	323	332	310.2
95%	330	341	322
dried	337	349	331.8
Flash point /°C	64	66	>70
Cetane number	74.1	66	75-90
Total aromatics %	0.4	0.7	-
Olefin /%	0.5	-	
Sulfur / ppm	< 4	<5	<0.5
CFPP /°C	-3	-34	-2

# Process integration research

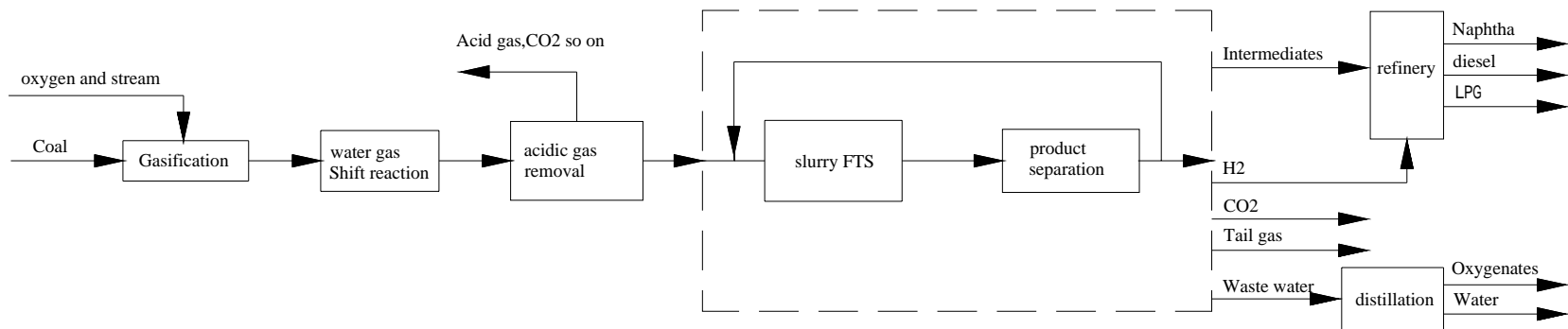
Coal gasification

Water gas shift reaction

Acidic gas removal

Slurry FTS

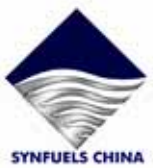
Product upgrading



## Comparison between simulation and industrial results

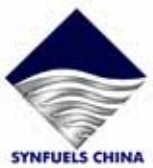
	Coal slurry coal gasification		GSP coal gasification	
	Industrial value	simulation	Industrial value	simulation
Gasification temperature	1400		1500	
Gasification pressure ( Mpa )	4.0		3.0	
Coal slurry concentration(wt%)	63.96%		-	
Coal conversion %	98		99.6	
syngas %(dry base)				
<b>H<sub>2</sub></b>	44.080	45.0227	27.06	27.06
<b>CO</b>	35.102	34.0537	68.36	68.28
<b>CO<sub>2</sub></b>	19.970	20.0943	4.06	4.18
<b>N<sub>2</sub></b>	0.374	0.5195	0.395	0.39
<b>NH<sub>3</sub></b>	0.134	0.0034	0.00	0.00
<b>H<sub>2</sub>S</b>	0.150	0.1526	0.04	0.04
<b>others</b>	0.08	0.0994	0	0

	SHELL gasification Illinois 6# coal		SHELL gasification Texas Lignite	
	literature	calculation	literature	calculation
Gasification temperature	1699.85		1644.25	
Gasification pressure ( Mpa )	2.308		2.7	
Coal conversion %	99.5		99.32	
syngas %(dry base)				
<b>CO</b>	64.38	65.57	62.69	63.80
<b>H2</b>	27.31	27.23	28.50	28.28
<b>CO2</b>	1.52	1.32	2.86	3.04
<b>CH4</b>	0.03	0.06	0.03	0.07
<b>Ar</b>	1.12	0.09	1.06	0.09
<b>N2</b>	4.17	3.11	4.49	4.27
<b>H2S</b>	1.33	1.26	0.33	0.41
<b>COS</b>	0.14	0.13	0.03	0.04



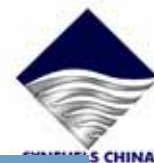
# Poly-generation : the future of CTL in China

- Poly-generation based on coal gasification is a promising sustainable strategy
- Poly-generation of CTL and co-produce electricity by combined cycle is an efficient way.
- The optimization of the poly-generation system should be carefully evaluated.



## Summary

- CTL in china can be a feasible way from viewpoint of technology, resource, economics, and environment
- CTL technology developed in Synfuels China has been tested in pilot plant, can be scaled up to demonstration plant scale. The demonstration plant is been design.
- Poly-generation of oil and electricity, and chemicals is the future of Large CTL



Thank you

